The field of high pressure X-rays, in particular, is developing so rapidly that this section is necessarily quite out of date by now.

In a very rough way, one could say that Volume 1 covers "physical" phenomena and Volume 2, "chemical" material. There is, however, much of interest to the physical chemist or chemical physicist in Volume 1, and the shock wave chapter in Volume 2 is of definite value to the physicist. The articles are largely self-contained, and a person interested in a particular area could readily buy and use one volume only.

The quality of figures and of printing is generally good. The one complaint the writer would make is concerning the variety of pressure units used. These include bars and kilobars atmospheres, dynes per sq. cm., and kilograms per sq. cm. This adds an entirely unnecessary confusion to the study of different chapters.

It is probably of interest to compare these volumes with the recent book, "Solids Under Pressure" (Paul and Warschauer, Editors). The treatment of solids and, in particular, physical phenomena in solids is much more thorough in the latter book, but, of course, the subject matter of the present volumes is considerably more comprehensive. The quality of the individual chapters is comparable.

There has been a considerable increase in interest in high pressure phenomena in the past fifteen years, stimulated in part by the synthesis of diamond, but in large part also by an increased understanding of the importance of interatomic distance as a parameter for understanding the physics and chemistry of matter. With the development of more and more very different techniques for studies to be made with vastly different purposes in mind, and to test theories with little or no overlap, it is becoming less and less feasible to consider "high pressure" per se as an intelligible area for a conference or a book. In the future, high pressure results should and will be included in conferences and books on cohesive energies, chemical kinetics, molecular spectroscopy, chemical synthesis, electronic structure, etc.

If, indeed, this is the swan song of the general and comprehensive book on high pressure, then it is ending on a very fine note.

DEPARTMENT OF CHEMISTRY AND CHEMICAL ENGINEERING UNIVERSITY OF ILLINOIS URBANA, ILLINOIS 61803 H. G. DRICKAMER

The Chemistry of Beryllium. Topics in Inorganic and General Chemistry. By D. A. EVEREST, The National Chemical Laboratory, Teddington, Great Britain. American Elsevier Publishing Co., Inc., 52 Vanderbilt Ave., New York 17, N. Y. 1964. 151 pp. 14 × 22 cm. Price, \$8.00.

This is the first of a projected series of monographs to appear under the collective title "Topics in Inorganic and General Chemistry." Dr. Everest has surveyed the chemistry of beryllium and presented it in a concise but readable book that is not too difficult for a junior or senior chemistry major. A teacher can find in it fresh and interesting examples to illustrate general chemical principles. The chemist who wishes to become acquainted with beryllium chemistry should start here although he will need to go to the original literature for more detail before beginning research in the field. The numerous references afford easy access to the literature, which appears to have been covered through early 1963.

After a brief historical introduction the behavior of the Be⁺² ion in aqueous solution is discussed with the emphasis on its hydrolysis. The preparation and behavior of the salts of oxyacids, the halides, and complexes (including the oxide carboxylates) are treated in the next three chapters. Two chapters on simple binary compounds other than the halides and on organoberyllium compounds end the review of beryllium compounds. The last four chapters cover extractive metallurgy, analytical chemistry, health hazards, and nuclear properties. Applications of beryllium compounds are not discussed except as examples. By relating and comparing the behavior of one compound or class of compounds with the others as he goes along the author largely avoids the choppy, disconnected effect that is so common in a compound by compound enumeration.

In view of the general level of the presentation, I think more explanation and interpretation of Diebler and Eigen's work on BeSO₄ solutions is needed; the equation on p. 11 is certainly not

clear. The phase equilibrium diagram for BeF_2 on p. 41 should have been simplified or else all of the features it shows should have been discussed. There are, I feel, some occasional infelicities of style that indicate the publisher did not exercise proper editorial care. There are few misprints, and these minor.

DEPARTMENT OF CHEMISTRY UNIVERSITY OF ILLINOIS URBANA, ILLINOIS R. F. TRIMBLE

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